

(Modern Physics)

Quiz 4 - PHY244

Print (HUGELY) Last Name: _____

Show detailed work to receive full scores.

First name: _____

Solution ^x

Giving or receiving aid is cause for dismissal from the university.

$$m_e c^2 = 511000 \text{ eV}, c = 3 \times 10^8 \text{ m/s}, hc = 1240 \text{ nm}\cdot\text{eV}, e = 1.6 \times 10^{-19} \text{ C}.$$

♠ An electron is confined in a one-dimensional box of a size a with a ground state energy 0.15 eV (relative to the bottom of the potential well). It jumps down from the first excited state to the ground state by emitting a single photon. Find the wavelength (in nm) of the photon.

$$E_1 = 0.15 \text{ eV}$$

$$E_2 = n^2 E_1 \quad \text{with } n=2 \\ = 4E_1 = 0.6 \text{ eV}$$

$$E_2 - E_1 = 0.45 \text{ eV}$$

$$\lambda = \frac{hc}{\Delta E} = \frac{1240 \text{ eV}\cdot\text{nm}}{0.45 \text{ eV}} = \underline{\underline{2756 \text{ nm}}}$$

♠ If the box size is reduced to $0.7a$ Find the photon wavelength (in nm) of the corresponding transition from the first excited state to the ground state.

$$E_1 \text{ becomes} = \frac{0.15 \text{ eV}}{0.7^2} = 0.3061 \text{ eV}$$

$$E_2 \text{ become} = \frac{0.6 \text{ eV}}{0.7^2} = 1.224 \text{ eV}$$

$$E_2 - E_1 = 0.918 \text{ eV}$$

$$\lambda = \frac{1240 \text{ eV}\cdot\text{nm}}{0.918} = \underline{\underline{1351 \text{ nm}}}$$